

Date Planned ://		Daily Tutorial Sheet-2	Expected Duration : 90 Min		
Actu	al Date of Attempt : / /	Level-1	Exact Duration :		
16.	Which one of the following is not a r	?			

16.	Which	Which one of the following is not a method of concentration of metals?									
	(A)	Gravity separation			(B)	Froth flotation process					
	(C)	Electromagnet	ic sepa	ration	(D)	Smelting					
17.	Froth floatation process used for the concentration of sulphide ore :										
	(A)	is based on the difference in wettability of different minerals									
	(B)	uses sodium ethyl xanthate, C ₂ H ₅ OCS ₂ Na as collector									
	(C)	uses NaCN as depressant in the mixture of ZnS and PbS when ZnS forms soluble complex and									
		PbS forms froth									
	(D)	all of the above are correct									
18.	Gravi	Gravity separation process is used for the concentration of :									
	(A)	calamine	(B)	haematite	(C)	chalcopyrites	(D)	bauxite			
19.	Froth floatation process for the concentration of sulphide ores is an illustration of the practical application of										
	(A)	Adsorption	(B)	Absorption	(C)	Sedimentation	(D)	Coagulation			
20.	Froth floatation process is used for the concentration of the ore of :										
	(A)	Fe	(B)	Al	(C)	Cr	(D)	Cu			
21.	Haem	Haematite ore is concentrated by :									
	(A)	Gravity separation method			(B)	Froth floatation	Froth floatation process				
	(C)	Amalgamation			(D)	Hand picking					
22.	Electr	Electromagnetic separation is used in the concentration of :									
	(A)	Copper pyrite	(B)	Bauxite	(C)	Cassiterite	(D)	Cinnabar			
23.	Which	Which one of the following is not a method of concentration of ore?									
	(A)	Electromagnetic separation			(B)	Smelting					
	(C)	Gravity separation			(D)	Froth floatation process					
24.	Chemical leaching is useful in the concentration of :										
	(A)	Copper pyrite	(B)	Bauxite	(C)	Cassiterite	(D)	Galena			
25.	The ore which is concentrated wetting by oil is :										
	(A)	oxide ore	(B)	sulphate ore	(C)	carbonate ore	(D)	sulphide ore			
26.	Magn	Magnetite is separated from chlorapatite by :									
	(A)	Froth floatation method			(B)	Levigation					
	(C)	Magnetic separation method			(D)	Electrostatic separation method					
27.	Which	Which of the following pair is incorrectly matched?									
	(A	Van Arkel method – Zirconium			(B)	Kroll's process – Titanium					
	(C)	Froth Floatation – Cerussite			(D)	Distillation – Zinc					



- 28. In the extraction of copper from its sulphide ore, the metal is formed by reduction of $\mathrm{Cu}_2\mathrm{O}$ with :
 - (A) FeS
- **(B)** C
- (C) Cu_2S
- SO_2

29. Out of the following reduction processes



 $\textbf{I.} \hspace{1cm} \text{Fe}_2 \text{O}_3 + \text{C} \longrightarrow \text{Fe}$

- $\textbf{II.} \qquad ZnO + C \longrightarrow Zn$
- III. $\operatorname{Ca}_3(\operatorname{PO}_4)_2 + \operatorname{C} \longrightarrow \operatorname{P}$
- **IV.** $PbO + C \longrightarrow Pb$

Correct processes are:

- (A) All of these
- (B) all but III
- (C) all but IV
- (**D**) II and IV
- **30.** Consider the following steps $CuS \xrightarrow{roast \text{ in air}} (A) \xrightarrow{roast \text{ without air}} (B)$



Which is not the correct statement?

- (A) It is self-reduction
- **(B)** It involves disproportionation $Cu_2S \longrightarrow Cu + CuS$
- (C) (A) is a mixture of CuO and CuS and (B) is a mixture of Cu and ${\rm SO}_2$
- **(D)** All are incorrect statements